# Monitoring Subsurface Fluid Flow using Perfluorocarbon Tracers: another tool potentially available for subsurface fluid flow assessments

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National Laboratory

### Candidate Tracers for Verification or Assessment (complementing geophysics)

**<u>Brines</u>**: Native non-conservative tracers that respond to changes pH, alkalinity, electrical conductivity Cations: Na, K, Ca, Mg,  $\Sigma$ Fe, Sr, Ba, Mn Major anions: CI, HCO<sub>3</sub>, SO<sub>4</sub>, F Organics: DOC, acetate, methane, benzene, toluene

<u>Gases</u>: Native conservative tracers or added conservative tracers
Ions: Br, I (*Na, K*) Gases: CO<sub>2</sub>, N<sub>2</sub>, H<sub>2</sub>, CH<sub>4</sub>, C<sub>2</sub> – C<sub>n</sub>
Noble gas tracers: Ar, Kr, Xe, Ne, He (and their isotopes)
Perfluorocarbon tracers (PFT's):
PMCP, PECH, PMCH, PDCH, PTCH (SF<sub>6</sub>)

<u>Isotopes</u>: D/H, <sup>18</sup>O/<sup>16</sup>O, <sup>87</sup>Sr/<sup>86</sup>Sr in water, DIC, minerals <sup>13</sup>C/<sup>12</sup>C in CH<sub>4</sub>, CO<sub>2</sub>, DIC, DOC, carbonate minerals

#### Perfluorocarbon Tracers (PFTs) Complement stable Isotopes and Geochemistry for Verifying, Assessing or Modeling Fluid Flow

PFTs areConservative, Non-reactive & Non- Hazardous tracers

PFT's sensitive at pg-fg, (versus stable isotopes at ppt)

PFT's easy and cheap as multiple combinations or suites for multiple breakthroughs

Complements geochemistry and geophysics providing multiple lines of evidence for flow path assessment

Applicable at near-surface or depth

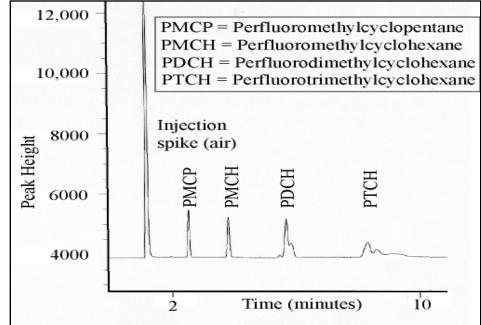
Scalable to thousands of samples

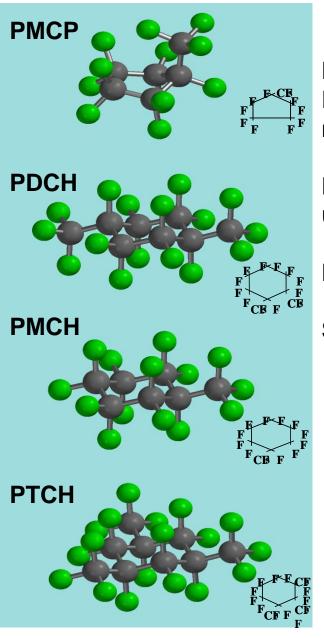
# capture detection

Proven established procedures

Analysis uses GC with electron

Can be analyzed in field or preserved





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### Example PFTs used and sample collection

**Deploy multiple-tracer suites (others available)** Different molecular weights, solubilities, and structure may enable chromatographic separation in reservoirs

Pressure cylinders for sample collection (U-tube) or use of serum vials that are inverted for storage

PFT Analyses performed in the field or preserved

Stable isotope analyses from pressurized samples





7-10 trucks/day Dayton Refinery

Fresh water (USDW) zone protected by surface casing

> Injection zones: 2004 experiment 2006 experiment

> Previous oil production Presentation\_

# **Brine Pilot Site**



Injection intervals: Oligocene fluvial and reworked sandstones:

• Porosity 34-24%,

South Salt

Liberty dome

- Permeability 4.4-2.5 Darcys,
- Steeply dipping 11 to 16°,
- Seals several thick shales,
- Depth 1,500 and 1,657 m,
- Brine-rock system150 and 165 bar
- Temperature 53 -60°C,
- Supercritical CO<sub>2.</sub>

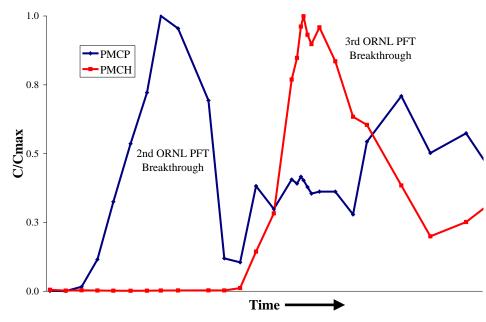
### PFT injection results

#### PFT Travel Time

- Travel time nearly constant (50 ±1.6 hr)
- Well developed CO<sub>2</sub> flow path

Peaks Broadened with time implying;

- PFTs were dispersing in the CO<sub>2</sub> throughout the experiment
- Flow paths continued to develop as the  $CO_2$  injection progressed



Injection #	Injection time (hours after CO <sub>2</sub> start)	Injection Duration (hours)	Peak Arrival Time (hours)	PFT Travel Time (hours) (GC)	PFT Travel Time (hours) (MS)	PFT Peak Broadness (hours) (GC and MS)
#1 PMCH/PTCH	2	4	54	50	49	14
#2 PMCP/PDCH	103	0.6	157	52	49	20
#3 PMCH/PTCH	120	0.5	173	51	53	24

#### Travel Times of Tracer Breakthroughs and Major Peaks (2<sup>nd</sup> site)

December			April			
Breakthrough/Maximum Peak		ak	Breakthrough / Major Peaks, Maximum			
		(Travel time in h	r after injection)			
		Monitoring Well	~ 50m			
PMCH	-/ 182		PMCP	288/ 360, <b>530,</b> 861		
PTCH	-/ 177		PDCH	288/ 359, <b>497</b> , 861		
Increased	Increased flow front 35/ 38 <sup>b</sup>		PECH	284/ 357, <b>423,</b> 446,810		
			SF6	284/ 370 <sup>a,</sup> <b>405,</b> 426, 841		
			PTCH/PMCH	>150/ *		
		Monitoring Well	~100m			
PMCH		-/ 238	PMCP	240/ 313, 470, <b>808</b>		
PTCH	214/ <b>277</b>		PDCH	262/ 327, 477, <b>793</b>		
Icreased flow front 140/ 158 <sup>b</sup>		3b	PECH	262/ <b>419,</b> 787, <b>880</b>		
			SF6	299/ 402, <b>803</b>		
			PTCH/PMCH	169/ 197 *		

In April 2010 tracers were added at the following hours: PMCP & PDCH = hr 1; PECH = hr 52; SF6 = hr 54; PTCH & PMCH = hr 693
Missed result due to U-tube issues.\*. Experiment ended at hr 906. SF6 peak was >10 times larger exhibiting larger and longer peaks.
b. After 30 days the flow into the formation was nearly doubled.

Lessons Learned for Technology Transfer Conduct base line characterizations before system is perturbed Utilize multiple chemical and isotopic probes and different suites of PFTs Deploy on-site analysis methods – e.g. pH, alkalinity Continue to monitor after test completion (surface and at depth) Integrate results with geophysics and coupled reactive-transport modeling PFTs cost < 1 cent per ton injectate (~ 0.1-1 ppm of fluid)

Summary: PFTs are Low cost, Non-toxic, Scalable, Sensitive (pg-fg;10<sup>-12-15</sup> quantities)

Geochemistry, Isotopes and PFT's complement Geophysics to monitor and verify plume movement, leakage to shallow aquifers or surface

PFTs: another tool available for potential leakage assessments

