

Objectives

- FTIR analysis of OA functional groups in SOAS samples.
- Lab studies of effects of VOCs, NO_x , and sulfate, on functional group composition and FTIR spectra of SOA formed from biogenic VOCs.
- Identify and quantify contributions of biogenic and anthropogenic emissions to SOA formed during SOAS using FTIR spectra plus lab and field results.



Laboratory Apparatus & Methods

Blacklights

Environmental Chamber [8 m³]

- VOC: isoprene, monoterpenes, alkanes (~0.1–1 ppmv)
- Seed particles: DOS, (NH₄)₂SO₄/H₂SO₄ (pH = 1, E-AIM)
- RH: 0 or 50% (SOAS ~ 75%)
- Oxidant & NO_x: OH (CH₂ONO/NO or H₂O₂ + UV), NO₃ (N₂O₅), O₃

Off-Line Analyses

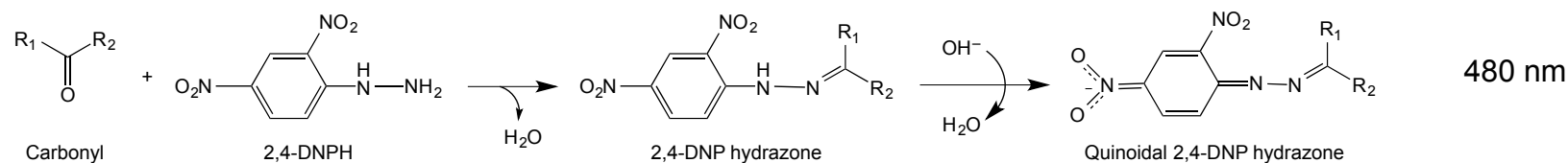
- Derivatization-Spectrophotometry, FTIR (extracts & filters) [particle functional groups]
- Gas & Liquid Chromatography & Mass Spectrometry [gas & particle composition]

On-Line Analyses

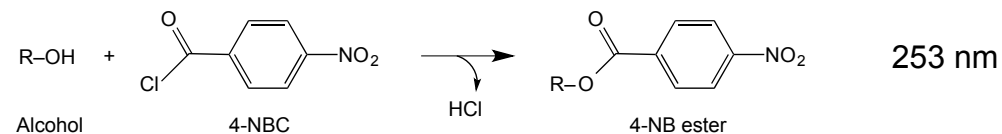
- Thermal Desorption Particle Beam Mass Spectrometer [particle composition & volatility]
- Scanning Mobility Particle Sizer [particle number and mass concentrations]

Derivatization-Spectrophotometric Methods for Functional Group Analysis

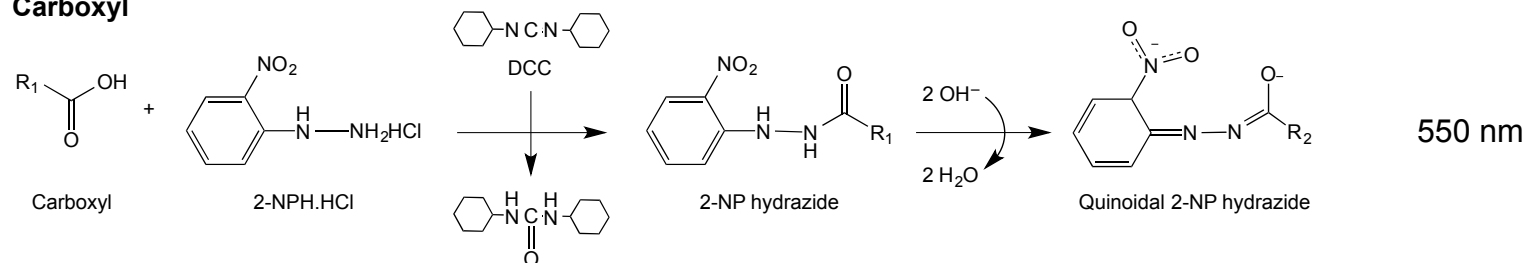
Carbonyl



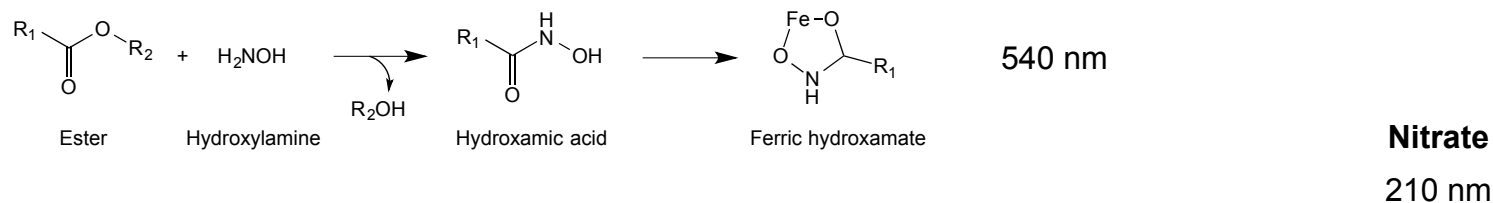
Hydroxyl



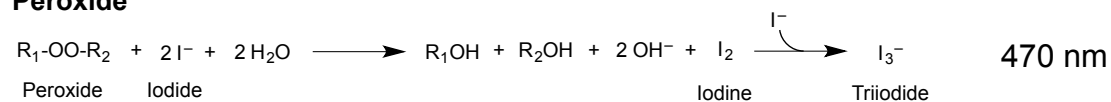
Carboxyl



Ester

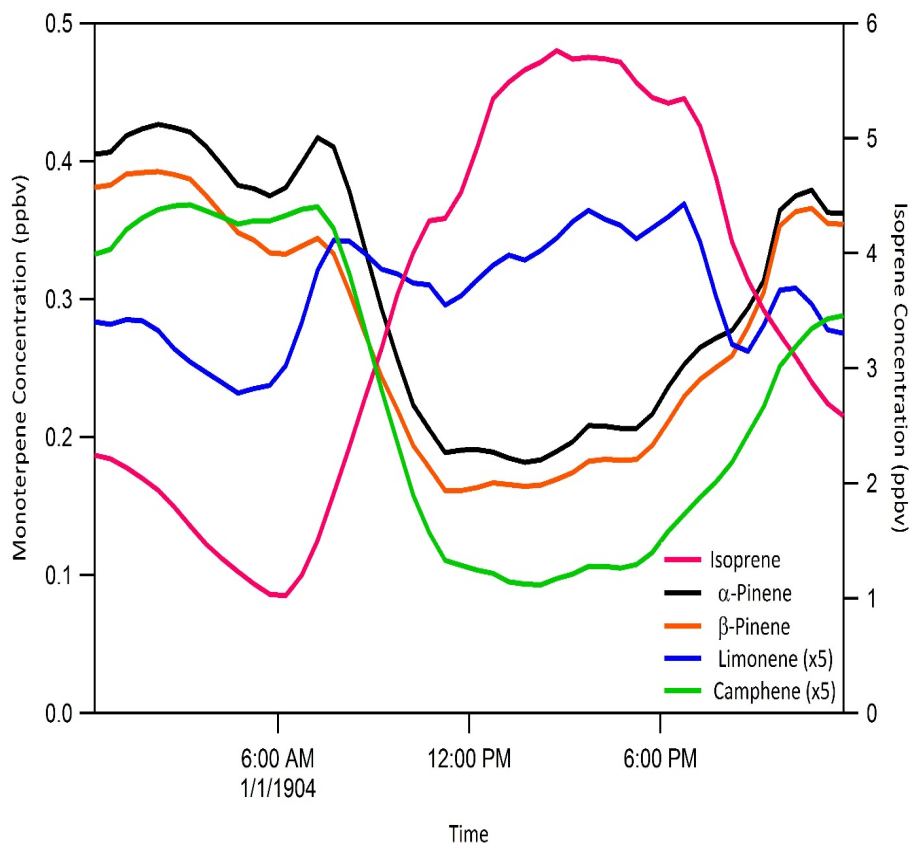




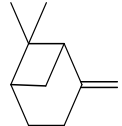
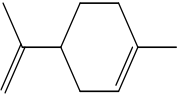
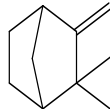
Peroxide



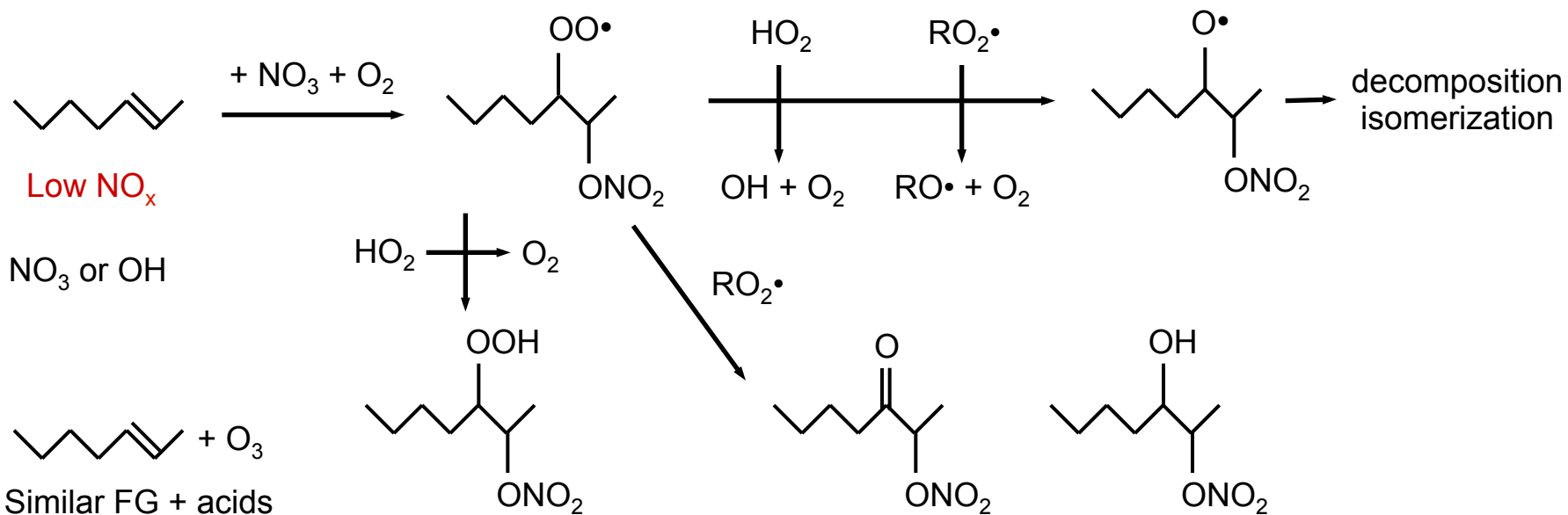
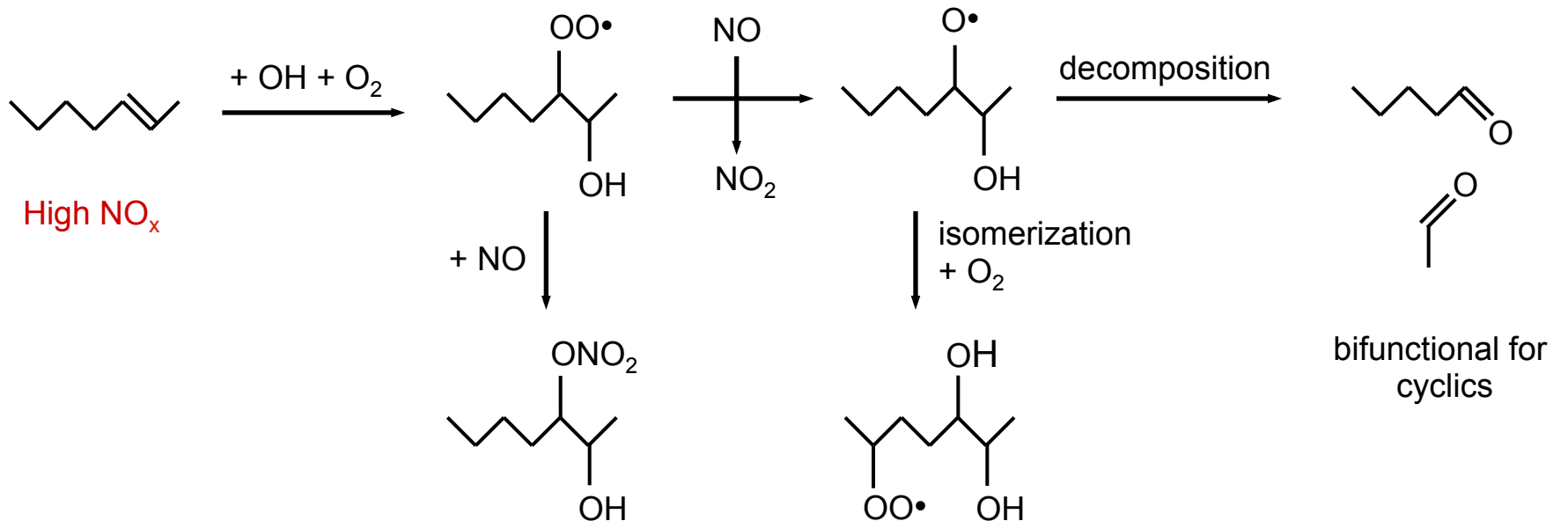
Selected VOCs

SOAS



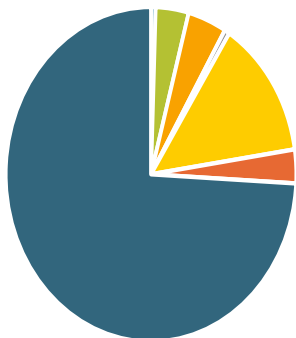
Compound	SOAS (ppbv)	Structure
Isoprene	3.5	
α -Pinene	0.31	
β -Pinene	0.28	
Limonene	0.06	
Camphene	0.05	

Reaction Mechanisms & Functional Groups



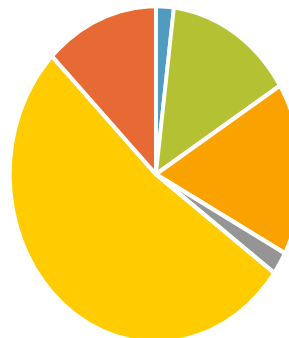
n-Pentadecane

OH/NO_x



with CH₂

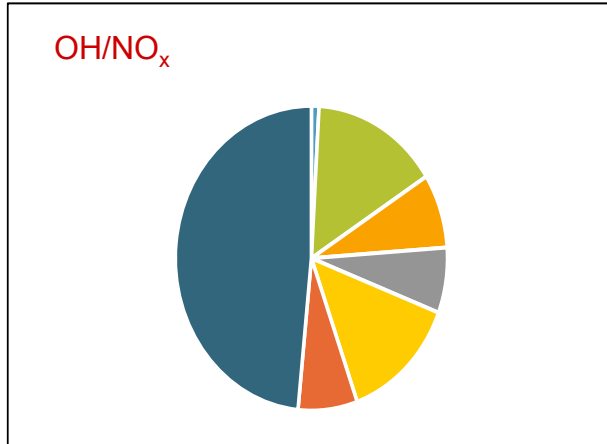
OH/NO_x



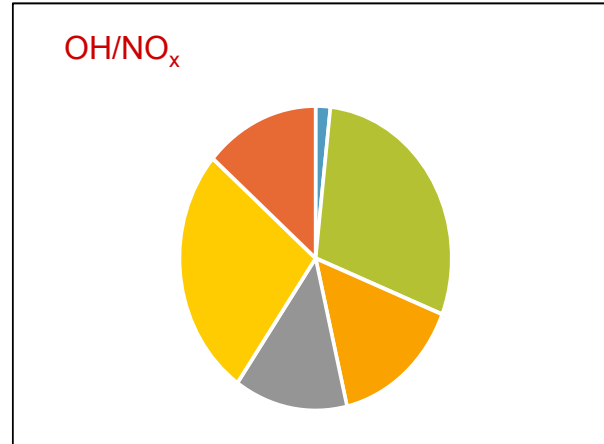
without CH₂

■ Peroxide ■ Ester ■ Carbonyl ■ Carboxyl ■ NO₃ ■ OH ■ CH₂

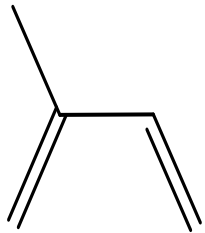
Isoprene



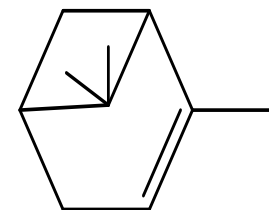
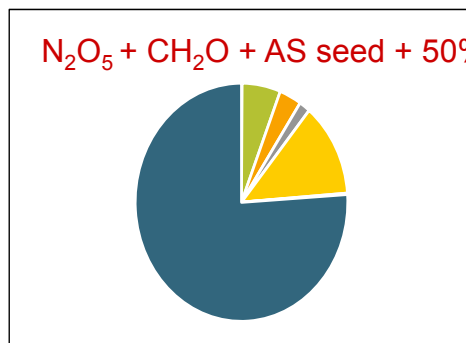
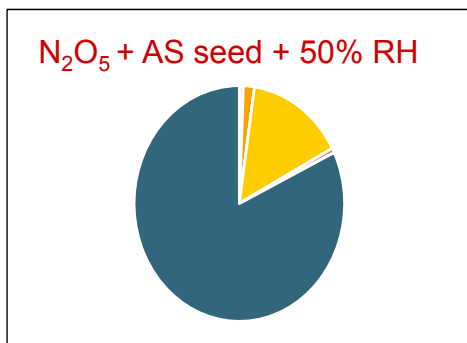
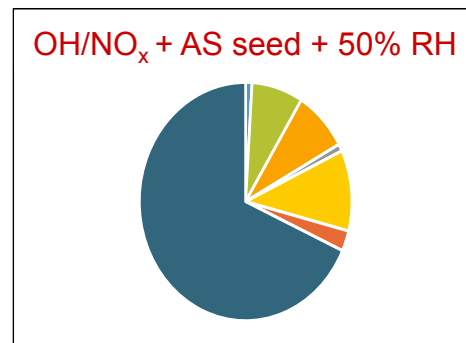
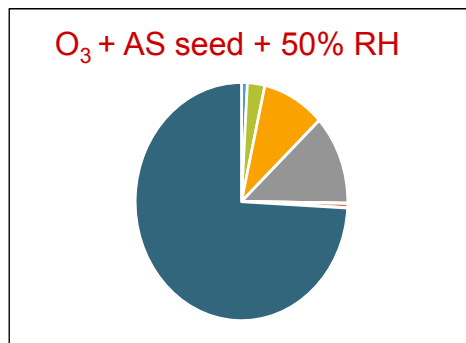
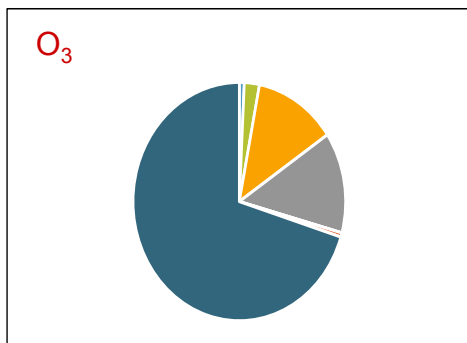
with CH₂



without CH₂



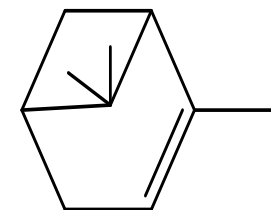
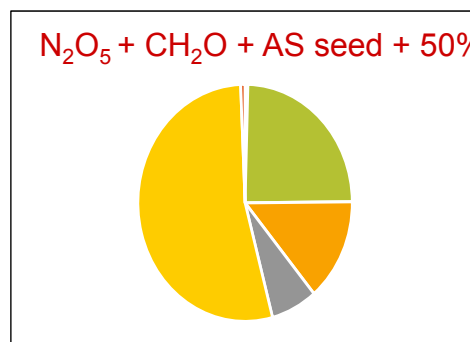
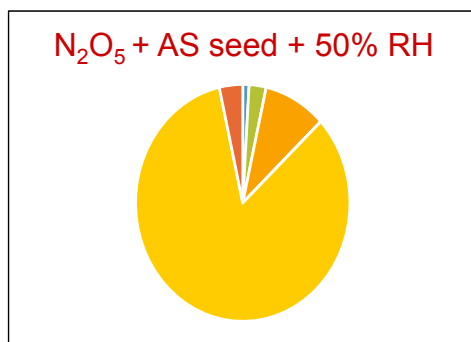
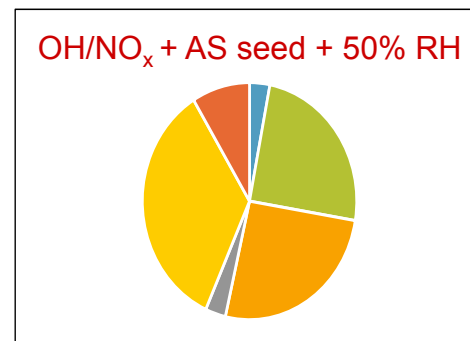
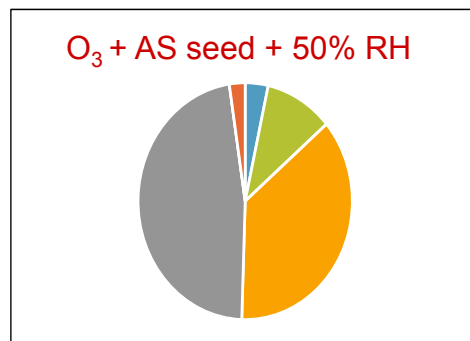
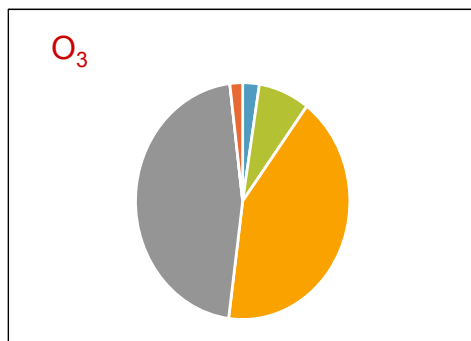
α -Pinene



with CH_2



α -Pinene

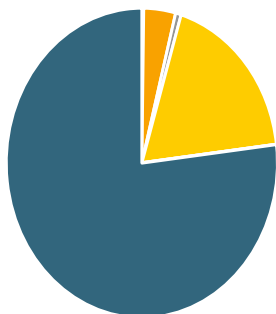


without CH_2

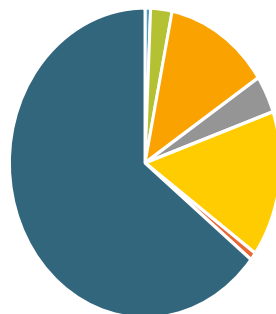


β -Pinene

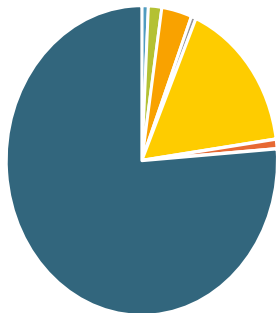
N_2O_5 + AS seed + 50% RH



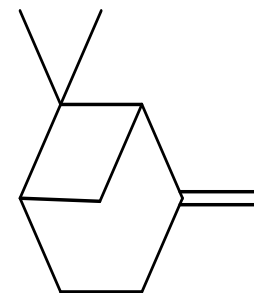
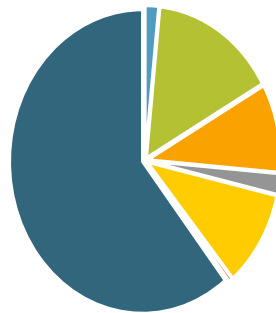
N_2O_5 + CH_2O + AS seed + 50% RH



N_2O_5 + H_2SO_4 /AS seed + 50% RH



N_2O_5 + CH_2O + H_2SO_4 /AS seed + 50% RH

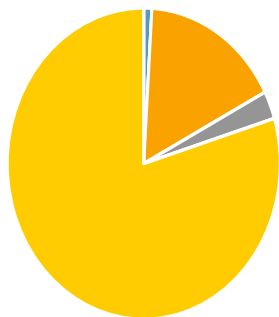


with CH_2

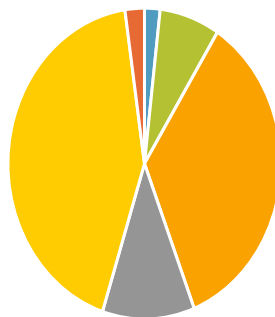
■ Peroxide ■ Ester ■ Carbonyl ■ Carboxyl ■ NO3 ■ OH ■ CH2

β -Pinene

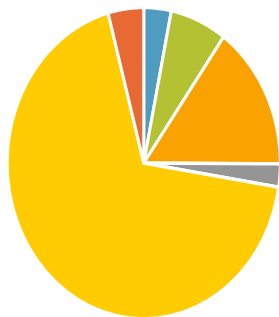
N_2O_5 + AS seed + 50% RH



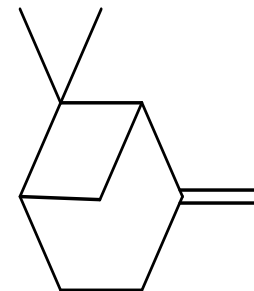
N_2O_5 + CH_2O + AS seed + 50% RH



N_2O_5 + H_2SO_4 /AS seed + 50% RH



N_2O_5 + CH_2O + H_2SO_4 /AS seed + 50% RH

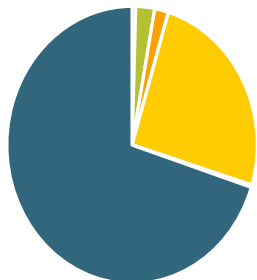


without CH_2

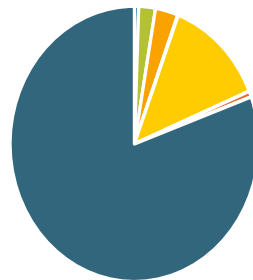
■ Peroxide ■ Ester ■ Carbonyl ■ Carboxyl ■ NO3 ■ OH ■ CH2

Limonene

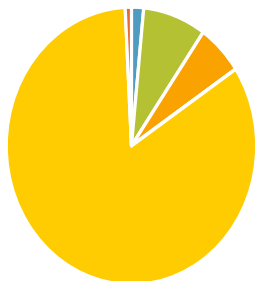
N_2O_5 + AS seed + 50% RH



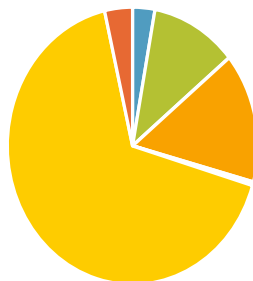
N_2O_5 + CH_2O + AS seed + 50% RH



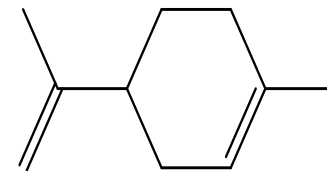
N_2O_5 + AS seed + 50% RH



N_2O_5 + CH_2O + AS seed + 50% RH



with CH_2

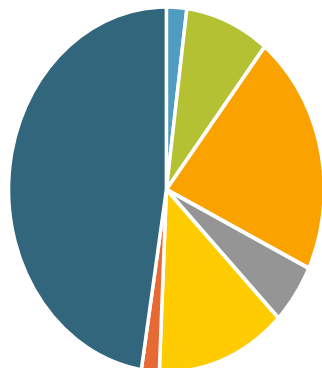


without CH_2

■ Peroxide ■ Ester ■ Carbonyl ■ Carboxyl ■ NO3 ■ OH ■ CH2

Camphene

$N_2O_5 + CH_2O + AS \text{ seed} + 50\% RH$

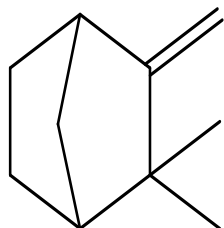


with CH_2

$N_2O_5 + CH_2O + AS \text{ seed} + 50\% RH$



without CH_2



■ Peroxide ■ Ester ■ Carbonyl ■ Carboxyl ■ NO3 ■ OH ■ CH2

Conclusions & Future Plans

- Observe all functional groups
- Clear difference between high and low NO_x compositions
- Experiments at lower VOC concentrations
- Experiments with variable $\text{HO}_2/\text{RO}_2\cdot$ ratios
- Experiments with mixtures
- Molecular analysis
- Analyze SOAS filter samples with derivatization spectrophotometric methods
- Compare FTIR and derivatization-spectrophotometric methods