

Cover Sheet for

**ENVIRONMENTAL CHEMISTRY METHOD**

***Pesticide Name:*** Cycloate

***MRID #:*** 415824-05

***Matrix:*** Soil

***Analysis:*** GC/NPD

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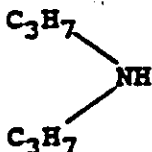
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**DETERMINATION OF DI-N-PROPYLAMINE, HEXAMETHYLENEIMINE AND  
ETHYLCYCLOHEXYLAMINE RESIDUES IN SOIL BY DIRECT  
EXTRACTION AND CAPILLARY GAS CHROMATOGRAPHY**

**I. SUMMARY/INTRODUCTION**

This method is intended for determining di-N-propylamine, hexamethyleneimine and ethylcyclohexylamine in soils at levels of 0.01 ppm to 0.5 ppm. Di-N-propylamine is a metabolite of EPTC, which is the active ingredient in EPTAM Selective Herbicide, manufactured by ICI Americas Inc. Their structures are as follows:

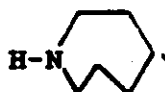


Di-N-Propylamine



Eptam

Hexamethyleneimine is a metabolite of molinate. Molinate is the active ingredient in ORDRAM Selective Herbicide, manufactured by ICI Americas Inc. Their structures are as follows:

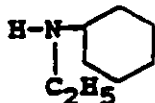


Hexamethyleneimine

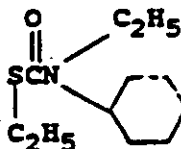


Molinate

N-ethylcyclohexylamine is a metabolite of cycloate. Cycloate is the active ingredient in RONEET Selective Herbicide, manufactured by ICI Americas Inc. Their structures are as follows:



N-Ethylcyclohexylamine



Cycloate

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RR90-409B

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## II. MATERIALS/METHODS

The equipment and reagents described below were used to generate the data and chromatograms presented in this report. Equipment with equivalent performance specifications and reagents of comparable purity can be used.

### A. Apparatus

1. Gas Chromatograph. Hewlett-Packard Model 5880A, equipped with capillary splitless inlet, Hewlett-Packard Model 7672A automatic sampler, nitrogen-phosphorus detector, and electronic integrator or data acquisition system. Any chromatograph giving equivalent performance may be used.
- ✓2. Injection Port Insert. Splitless insert, 2 mm i.d. by 77 mm, Hewlett-Packard Part No. 18740-80220.
- ✓3. Chromatographic Column. DB-5 (crosslinked 5% phenylmethyl silicone), 12 m x 0.2 mm x 0.33 μm thickness, or equivalent.
- ✓4. Glass Bottles. Four-ounce, wide mouth bottles with teflon lined caps, two-ounce narrow mouth bottles with plastic caps, and 4-dram vials with plastic caps.
- ✓5. Syringe. 10, 100, and 500 microliter capacities, Hamilton 701N, 710N, 750N or equivalent.
6. Reciprocating Shaker. Eberback Corporation, Model 6010 or equivalent.
7. Centrifuge. IEC International, Model C1582 or equivalent.

### B. Reagents

- ✓1. Solvents. Toluene, Acetone, Nanograde or equivalent.
2. Di-N-propylamine, hexamethyleneimine, N-ethylcyclohexylamine. Analytical reference standard available from ICI Americas Inc., 1200 South 47th Street, Box 4023, Richmond, CA 94804-0023, Attention: Environmental Sciences Department Manager.
3. Acetic Anhydride, NaCl, Anhydrous Na<sub>2</sub>SO<sub>4</sub>, K<sub>2</sub>CO<sub>3</sub>, 50% NaOH. Reagent grade or equivalent.
4. Calibration and Fortification Solutions

To prepare a stock solution of di-N-propylamine, hexamethyleneimine, or N-ethylcyclohexylamine, weigh to the 4th decimal place a convenient quantity, e.g.

approximately 50 mg, of primary standard of known purity into a suitably sized bottle. Calculate the weight of solvent to add, based on the weight of primary standard taken, the purity of the primary standard, the density of the solvent, and the desired solution concentration, typically 1000 µg/mL, as follows:

$$S = \frac{W \times P \times D}{A}$$

where S = the weight of solvent to add (g),

W = the weight of primary standard taken (mg std),

P = the purity of the primary standard (mg a.i./mg std),

D = the density of the solvent (g/mL),

and A = desired solution concentration (mg a.i./mL solvent).

Add the calculated weight of the appropriate solvent to the bottle, close the bottle with a polyseal cap, and mix thoroughly to dissolve the primary standard. Use toluene (D = 0.867 g/mL) for calibration solutions, and acetone (D = 0.792 g/mL) for fortification solutions.

To prepare working calibration solutions, dilute the stock calibration solution by weight with toluene to give 1.0, 0.1, and 0.02 µg/mL solutions or other concentrations as required.

Dilute the stock fortification solution by weight with acetone to give a 10 µg/mL solution, or other concentrations as required.

### C. Analytical Procedure

#### 1. Extraction

Weigh 40 g of thoroughly mixed soil sample into a 4-oz. wide-mouth bottle. Add 40 mL of distilled water and 10 g of NaCl. Then, add 5 mL of 50% NaOH and 20 mL of toluene. Cap the bottle with a teflon-lined lid and shake it for 2 hours. Centrifuge for 10 to 20 minutes at 2000 rpm to aid in the separation of phases. Remove the top (toluene) phase for derivitization. Alternatively use any convenient weight of soil >20 g, and extract with the same proportions (e.g. 20 g soil: 20 mL water: 5 g NaCl: 2.5 mL NaOH: 10 mL toluene).

**2. Derivative Formation**

Transfer 15 to 20 mL of the top layer of toluene from the centrifuged sample into a 4-ounce jar and dry it with about 5 g of granular anhydrous  $\text{Na}_2\text{SO}_4$ . Add 1 mL of reagent grade acetic anhydride (99.9%) and let stand, capped, one hour or more at room temperature. Then, transfer approximately 2 mL into a 4-dram vial, add an equivalent volume of 1 M  $\text{K}_2\text{CO}_3$ , cap the vial and shake it vigorously. Let the layers separate for about 5 minutes, transfer the top (toluene) into a separate 4-dram vial, and add about 2 g of anhydrous  $\text{Na}_2\text{SO}_4$ . This extracted and derivatized sample is injected into the G.C.

**3. Fortification**

Analyze unfortified and fortified control samples with each set of treated samples to demonstrate method recovery according to the Quality Assurance SOP. For example, for 40-g samples, weigh 40 grams of untreated control soil into a 4-oz wide mouth bottle. Add 0.040 mL of 10  $\mu\text{g}/\text{mL}$  acetone fortification solution (0.4  $\mu\text{g}$ ) to produce a fortification level of 0.01 ppm, or add 0.20 mL of 100  $\mu\text{g}/\text{mL}$  acetone fortification solution (20  $\mu\text{g}$ ) to produce a fortification level of 0.50 ppm. Add 40 mL of water, 5 g of NaCl, 5 mL of 50% NaOH, 20 mL of toluene and extract and derivatize as above. If a different weight of soil is analyzed, use that weight and adjust the volume or concentration of fortification solution to give the desired analyte concentration. Extract using the same volumes of water and toluene as for the treated samples.

**D. Instrumentation****1. Operating Conditions**

Follow the manufacturer's instructions for operation of the gas chromatograph and nitrogen-selective detector. Use these parameters for the analyses or other operating conditions that achieve equivalent sensitivity, reproducibility, and resolution.

Inlet	Splitless insert, purge activated at 0.5 min.
Oven initial temp.	90°C
Initial time	1.0 min
Temp. programming rate	15°C/min
Oven final time	10 min
Oven final temperature	220°C
Injector temperature	220°C
Detector temperature	280°C

Carrier gas	Helium
Carrier gas pressure	23 psi
Carrier gas flow	2.4 mL/min through column, 78 mL/min vented
Injection size	1.5 $\mu$ L
Quantitation	Peak height or area (external standard)

Under the above conditions the elution times of the analytes are: di-N-propylamine, 1.2 minutes; hexamethyleneimine 2.0 minutes; and N-ethylcyclohexylamine 2.8 minutes.

## 2. Calibration

The gas chromatograph is calibrated using the analyte calibration solutions specified in section II.B.3. Chromatographic sensitivity is established by analysis of the 0.02  $\mu$ g/mL calibration solution. Quantitation of residues at levels above the detection limit is done by an external standard procedure in which peak heights or areas of analyte peaks in sample extracts are compared to corresponding peak heights or areas of analyte peaks in calibration solutions. See Section G below for details of calculational methods.

## 3. Analysis of Extracts

Inject the sample extracts using the same conditions used for calibration. The identity of the analyte peak in the sample chromatogram is assigned based upon the coincidence of retention times (within 0.03 minutes) with those of the calibration chromatograms. If the response of a peak identified as an analyte exceeds that of the highest calibration solution, dilute the sample extract until its response is within the calibrated range. Reinject calibration solution after every two to four sample injections and recalibrate as needed. Reinject calibration solution at completion of the sample analysis.

## E. Interferences

No clean-up is required when this procedure is utilized as described. However, extractives from soil occasionally contribute peaks with retention times near those of the analytes. Satisfactory resolution can usually be achieved with appropriate oven temperature manipulations or column choice. Appendix A shows typical chromatograms. Analyze extracts of samples from untreated plots to demonstrate the absence of interferences from sample matrices, solvents, or labware. The chromatograms in Appendix A demonstrate resolution of di-N-propylamine, hexamethyleneimine and N-ethylcyclohexylamine.

**F. Confirmatory Techniques**

Unexpected positive results, as in untreated control or pre-application samples, should be confirmed by other means, preferably by GC/MS, mass selective detection, or use of a second capillary column of different polarity.

**G. Calculations**

Calculations are done in one of two ways. If the response is linear, a factor can be calculated as described in 1 below. If the response is non-linear, or if the analyst prefers, the analyte responses over a range of calibration solution concentrations can be fit to a linear or an exponential curve, and a factor can then be calculated as in 2 below for each point on the curve that corresponds to an analyte response in an injection of sample extract.

**1. Linear Response. Direct Calculation of Factor****a. Calibration Factors for Linear Response**

F = the response factor for the analyte (ppm per electronic unit), calculated as follows:

$$F = \frac{C}{P \times S}$$

where C = the concentration of analyte in the calibration solution ( $\mu\text{g/mL}$ )

S = the amount of initial sample represented by each milliliter of final extract solution injected ( $\text{g/mL}$ )

and P = the peak area or height (electronic units) of the analyte peak in the chromatogram of the calibration solution

Averaged response factors for multiple injections of calibration solutions and for more than one concentration of calibration solution can be used as appropriate in the calculation of the concentration of the analyte in the sample, as described below.

**b. Analyte in Sample**

The concentration of the analyte in the original sample is calculated using an external standard method as follows:

$$\text{ppm} = F \times R$$



where ppm = the amount of analyte in the soil in parts per million

R = the peak area or height (electronic units) of the analyte peak in the chromatogram of the sample extract

and F = the response factor for the analyte (ppm per electronic unit), calculated as described above

Note for the above external standard calculations, equal volumes of both the extract and the calibration solutions are injected.

✓ 2. Curve Fit for Linear or Non-Linear Response

If the instrumental response to injections of calibration solutions is reproducible and either linear or exponentially non-linear, a concentration-response curve can be used for sample quantitation. Any valid curve-fitting program can be used. Input the concentration and response for each injection of calibration solution. The program will generate the formula for the corresponding linear or exponential curve. From the formula, determine the calculated concentration for each injection of calibration solution as described below. The calculated and actual concentrations should agree within 10% relative; that is, the ratio of the actual to the calculated concentration should be between .9 and 1.1. If the agreement is adequate, calculate the concentration of analyte in the sample, and corresponding response factor as follows:

a. Linear Response:

The formula will be of form  $Y = mX + b$ , where

Y = the concentration of the analyte, ppm,

X = the analyte response, peak height or area units,

and

m and b = constants calculated by the curve fit program.

Since the analyte concentration should be zero if the response is zero, the constant b should be zero if there are no systematic errors in the analysis. However, it is not necessary for b to be zero for

the calculational method to be valid, as long as calibration solution responses are reproducible and the calculated concentrations of the calibration solutions are within 10% of the actual concentrations.

For each sample injection, determine Y by using the response, X, in the formula.

Calculate the response factor, F, from the formula:

$$F = Y/X$$

Note that this factor should be the same for any point on a linear curve which passes through the intercept;  $b = 0$ .

b. Exponential non-linear response:

The curve will be of form  $Y = aX^b$ , where

Y = the concentration of the analyte, ppm,

X = the analyte response, peak height or area units,

and

a and b = constants calculated by the curve-fit program.

For each sample injection, determine Y by using the response, X, in the formula.

Calculate the response factor, F, from the formula:

$$F = Y/X$$

The response factor will be different for each point on the curve.

### III. DISCUSSION

#### A. Precision and Accuracy

Fortified soil samples were prepared as described as under C-2 and analyzed according to the method to establish precision and accuracy of the method. The precision of this method depends on variations in extraction and instrumental analysis. The variations in extraction and instrumental analyses can be evaluated from the data obtained during analyses of fortified samples.

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## 1. Instrumental precision

Each extract of the fortified samples was analyzed three to six times to provide information on instrumental precision. For triplicate analyses, the precision is expressed as relative variance; that is, 100 times the maximum difference between an individual result and the mean value, divided by the mean value. Relative variance is used because the number of analyses is not large enough for the statistical variance,  $s$ , to be meaningful. Variance, defined below, is used in the calculation of the coefficient of variation,  $100s/x$ , which is used to describe precision when five or more analyses were performed (Ref. 2).

$$s^2 = \frac{\sum X_i^2 - \frac{1}{n} (\sum X_i)^2}{n-1}$$

Di-N-Propylamine

Table I shows relative variance for three injections of extracts. For samples fortified with 20 ug of analyte, extract concentration approximately 1 ug/ml, the relative variance ranged from 2.1 to 6.3 percent, with a mean of 4.2 percent. For samples fortified with 0.4 ug of analyte, extract concentration approximately 0.02 ug/ml, the variances were expectedly larger, ranging from 9.6 to 17 percent, with a mean of 12 percent.

Hexamethyleneimine

Table 2 shown a range of relative variance for three injections of extract from samples fortified with 20 ug of 4.6 to 13 percent, with a mean of 7.6 percent. For samples fortified with 0.4 ug, the range of coefficient of variation for 6 injections of extract was 18 to 32 percent, with a mean of 25 percent. This value, although relatively large, is acceptable of soil at concentrations of analyte near the detection limit of 0.01 ppm.

N-Ethylcyclohexylamine

Table 3 shows that the relative variance for three injections of extract from samples fortified with 20 ug of analyte ranged from 4.2 to 9.6 percent with a mean of 6.6 percent. The coefficient of variation of six injections of extract from samples fortified with 0.4 ug of analyte ranged from 8.7 percent to 21 percent, with a mean of 14 percent; again, this variability is acceptable for soil analyses at the detection limit.

## 2. Overall precision and accuracy

✓ Di-N-Propylamine

Table I shows that the mean recovery from five samples fortified with 0.40  $\mu\text{g}$  of di-N-propylamine was 0.451  $\mu\text{g}$  or 90 percent. The coefficient of variation from the five analyses was 4.1%. For six samples fortified with 20  $\mu\text{g}$ , the mean recovery was 19.8  $\mu\text{g}$  or 99%, and the coefficient of variation was 2.9%.

Hexamethyleneimine

Table 2 shows a mean recovery of 15.5  $\mu\text{g}$ , or 78%, for six samples fortified with 20  $\mu\text{g}$ ; the coefficient of variation was 3.0%. At a fortification level of 0.40  $\mu\text{g}$  five samples were analyzed; the mean recovery was 0.415  $\mu\text{g}$  (104%), and the coefficient of variation was 3.6%.

N-Ethylcyclohexylamine:

Table 3 shows that the mean recovery from five samples fortified with 0.40  $\mu\text{g}$  of N-ethylcyclohexylamine was 0.340  $\mu\text{g}$  or 85 percent. The coefficient of variation from the five analyses was 4.3%. For six samples fortified with 20  $\mu\text{g}$ , the mean recovery was 19.8  $\mu\text{g}$  or 99%, and the coefficient of variation was 5.9%.

B. Limit of Detection

The detection limit of the method is 0.01 ppm as determined by fortifications at the 0.01 ppm level with 2-7 cm peak heights.

C. Dry Weight Basis

This method determines the residues on an as-received basis. If it is desired to express the values on a dry-weight basis, compensation is necessary for water present in the sample. Percent moisture can be determined by drying a subsample at 105°C for 24 hours.

D. Safety Precautions

Personnel untrained in the routine safe-handling of chemicals and good laboratory practices should not attempt to use this procedure. Information on any specific chemical regarding physical properties, hazards, toxicity, and first-aid procedures can be found on the Material Safety Data Sheets accompanying the chemical or available from the

chemical supplier. In general, always wear safety glasses, work in a well ventilated area, avoid inhaling vapors, and avoid contact of any chemical with skin and clothing. Flammable solvents should be kept away from potential sources of ignition.

Toluene, Acetone

Flammable.

Avoid contact with skin and clothing.

Avoid breathing vapor; work in well ventilated area.

Di-n-propylamine, Hexamethyleneimine, N-ethylcyclohexylamine

Avoid contact with skin and clothing.

Work in well ventilated area.

Wash with soap and water after any accidental contact.

Acetic Anhydride, 50% NaOH

Corrosive.

Avoid contact with skin and clothing.

Rinse copiously after any accidental contact.

IV. CONCLUSIONS

The method is specific for the analysis of di-N-propylamine, hexamethyleneimine, and N-ethylcyclohexylamine in soil. Only readily available laboratory equipment and reagents are required. The analysis can be completed by one person in an 8-hour period if an adequately homogenized sample is available. Untreated and fortified untreated samples should be extracted and analyzed with each set of samples to demonstrate absence of interferences and adequate recovery. If determination of di-N-propylamine, hexamethyleneimine, or N-ethylcyclohexylamine at a concentration other than 0.01 ppm to 0.5 ppm is required, suitably fortified samples must be analyzed to validate the method at that concentration.

V. CERTIFICATION

This is to certify that this is a complete and unaltered report prepared by the Analytical Department of ICI Americas Inc., Western Research Center.

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APPROVALS

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5/12/89  
Date

Table 1. Recoveries from Soil Fortified with 0.40  $\mu\text{g}$  (0.01 ppm) and with 20  $\mu\text{g}$  (0.50 ppm) of Di-N-Propylamine

Sample Number	Fortification ( $\mu\text{g}$ )	$\mu\text{g}/\text{Found}$			Mean	RV <sup>1</sup>	Percent Recovery
		Run 1	Run 2	Run 3			
11808-36-2	0.4 <sup>0.01</sup>	0.450	0.462	0.406	0.439	7.6	88
11808-36-3	0.4	0.540	0.450	0.449	0.479	13	96
11808-36-4	0.4	0.376	0.488	0.499	0.454	17	91
11808-36-5	0.4	0.390	0.462	0.442	0.431	9.6	86
11808-36-6	0.4	0.390	0.514	0.449	0.451	14	90
				MEAN	0.451	12	90
				CV <sup>2</sup>	4.1		
11808-35-2	20 <sup>0.5</sup>	24.4	21.6	16.8	20.9	6.3	105
11808-35-3	20	19.3	20.6	18.2	19.4	4.3	97
11808-35-4	20	20.2	19.7	18.7	19.5	4.2	98
11808-35-5	20	20.8	19.3	19.8	20.0	3.0	100
11808-35-6	20	20.0	18.9	19.4	19.4	2.1	97
11808-35-7	20	20.0	20.0	20.0	19.8	4.0	92
				MEAN	19.8	4.2	99
				CV	2.9		

<sup>1</sup>RV = Relative Variance:  $100 (X-\bar{X})/\bar{X}$

<sup>2</sup>CV = Coefficient of Variation:  $100 S/\bar{X}$

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Table 2. Recoveries from Soil Fortified With 0.4  $\mu\text{g}$  (0.01 ppm) and With 20  $\mu\text{g}$  (0.50 ppm) of Hexamethylenimine

Sample Number	Fortification		$\mu\text{g}$ Found						Mean	CV <sup>1</sup>	Recovery
	( $\mu\text{g}$ )	(ppm)	Run 1	Run 2	Run 3	Run 4	Run 5	Run 6			
11808-36-2	0.4	0.01	0.284	0.324	0.320	0.451	0.471	0.508	0.393	24	98
11808-36-3	0.4	0.01	0.382	0.279	0.265	0.511	0.471	0.574	0.410	32	103
11808-36-4	0.4	0.01	0.444	0.337	0.322	0.481	0.471	0.492	0.425	18	106
11808-36-5	0.4	0.01	0.346	0.383	0.286	0.496	0.502	0.492	0.417	22	104
11808-36-6	0.4	0.01	0.395	0.377	0.233	0.496	0.549	0.541	0.432	28	108
MEAN									0.415	25	104
CV									3.6		

Sample Number	Fortification		$\mu\text{g}$ Found				Mean	RV <sup>2</sup>	Percent Recovery
	( $\mu\text{g}$ )	(ppm)	Run 1	Run 2	Run 3	Run 4			
11808-35-2	20	0.50	16.8	15.4	15.1	15.8	6.6	79	
11808-35-3	20	0.50	16.3	12.8	15.2	14.8	13	74	
11808-35-4	20	0.50	16.3	14.7	14.8	15.3	6.8	76	
11808-35-5	20	0.50	16.2	16.4	15.2	15.9	4.6	80	
11808-35-6	20	0.50	16.4	15.4	14.3	15.4	6.9	77	
11808-35-7	20	0.50	17.2	15.2	15.5	16.0	7.7	80	
MEAN							15.5	7.6	78
CV							3.0		

<sup>1</sup>CV = Coefficient of Variation:  $100 \frac{s}{\bar{x}}$

<sup>2</sup>RV = Relative Variance:  $100 \frac{(x-\bar{x})^2}{\bar{x}}$



**Table 3.** Recoveries from Soil Fortified with 0.4  $\mu\text{g}$  (0.01 ppm) and with 20  $\mu\text{g}$  (0.50 ppm) of Ethylcyclohexylamine

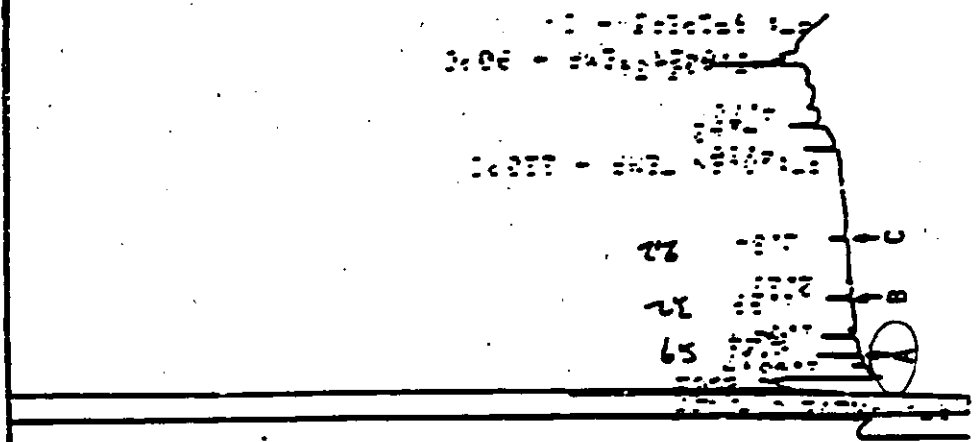
Sample Fortification Number	Fortification ( $\mu\text{g}$ )	$\mu\text{g}$ Found						Mean	CV <sup>1</sup>	Percent Recovery
		Run 1	Run 2	Run 3	Run 4	Run 5	Run 6			
11808-36-2	0.40	0.280	0.353	0.325	0.344	0.376	0.344	0.337	10	84
11808-36-3	0.40	0.328	0.265	0.263	0.407	0.391	0.344	0.333	18	83
11808-36-4	0.40	0.444	0.327	0.322	0.391	0.344	0.344	0.362	13	91
11808-36-5	0.40	0.294	0.241	0.266	0.391	0.407	0.344	0.326	21	81
11808-36-6	0.40	.296	0.306	0.321	0.360	0.345	0.344	0.345	8.7	86
MEAN								0.340	14	85
CV								4.3		

Sample Fortification Number	Fortification ( $\mu\text{g}$ )	$\mu\text{g}$ Found				Mean	RV <sup>2</sup>	Percent Recovery
		Run 1	Run 2	Run 3	Run 4			
11808-35-2	20	20.5	23.0	20.4	21.3	4.2	107	
11808-35-3	20	19.7	18.4	19.9	19.3	5.0	97	
11808-35-4	20	22.4	20.1	18.8	20.4	9.6	102	
11808-35-5	20	22.0	19.9	19.3	20.4	7.8	102	
11808-35-6	20	19.4	18.0	14.5	18.0	8.2	90	
11808-35-7	20	20.4	19.1	18.5	19.3	5.2	97	
MEAN					19.8	6.6	99	
CV					5.9			

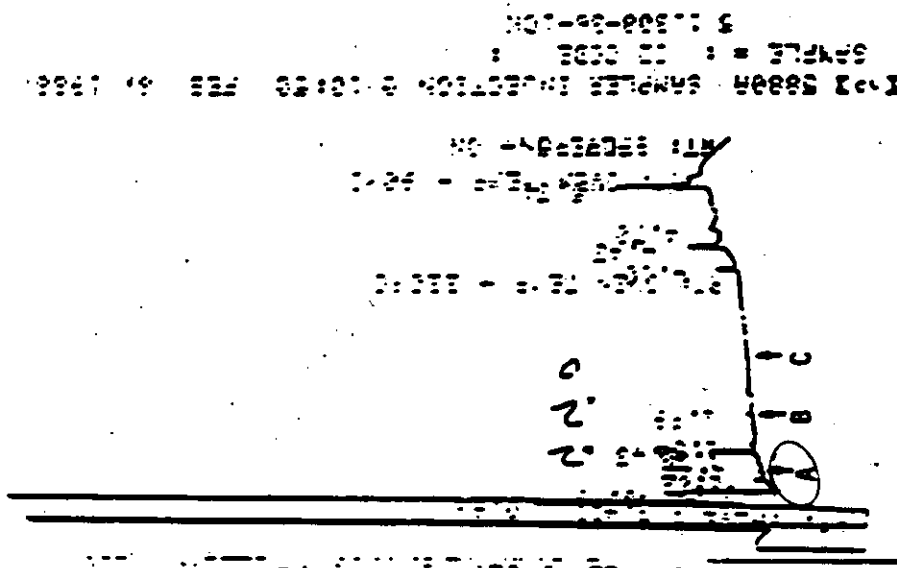
<sup>1</sup>CV = Coefficient of Variations:  $100 \frac{s}{\bar{x}}$

<sup>2</sup>RV = Relative Variance:  $100 \frac{(x-x)^2}{\bar{x}}$

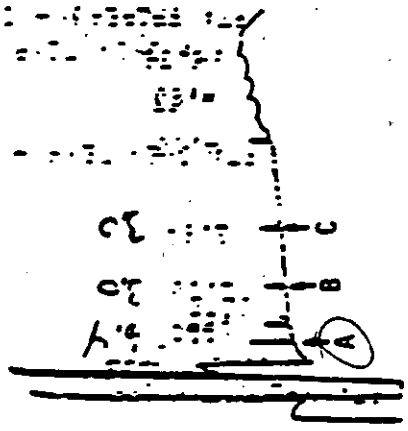
TABLE 3. RECOVERIES FROM SOIL FORTIFIED WITH 0.4  $\mu\text{g}$  (0.01 ppm) AND WITH 20  $\mu\text{g}$  (0.50 ppm) OF ETHYLCYCLOHEXYLAMINE



Untreated Control  
Fortified at 0.01 ppm



Untreated Control



Calibration Solution  
0.01 ug/ml

Figure 1a. Example chromatograms for the determination of (dl-n-propylamine (A) hexamethylenimine (B), and N-ethylcyclohexylamine (C) in soil; toluene extracts 2 g soil per mL toluene.

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VII. REFERENCES

1. WRC Laboratory Notebook 11235, pages 26-34.
2. Dixon, W. J. and F. J. Masey, Introduction to Statistical Analysis, McGraw-Hill Book Company, 1969, pp 27, 28, 113.

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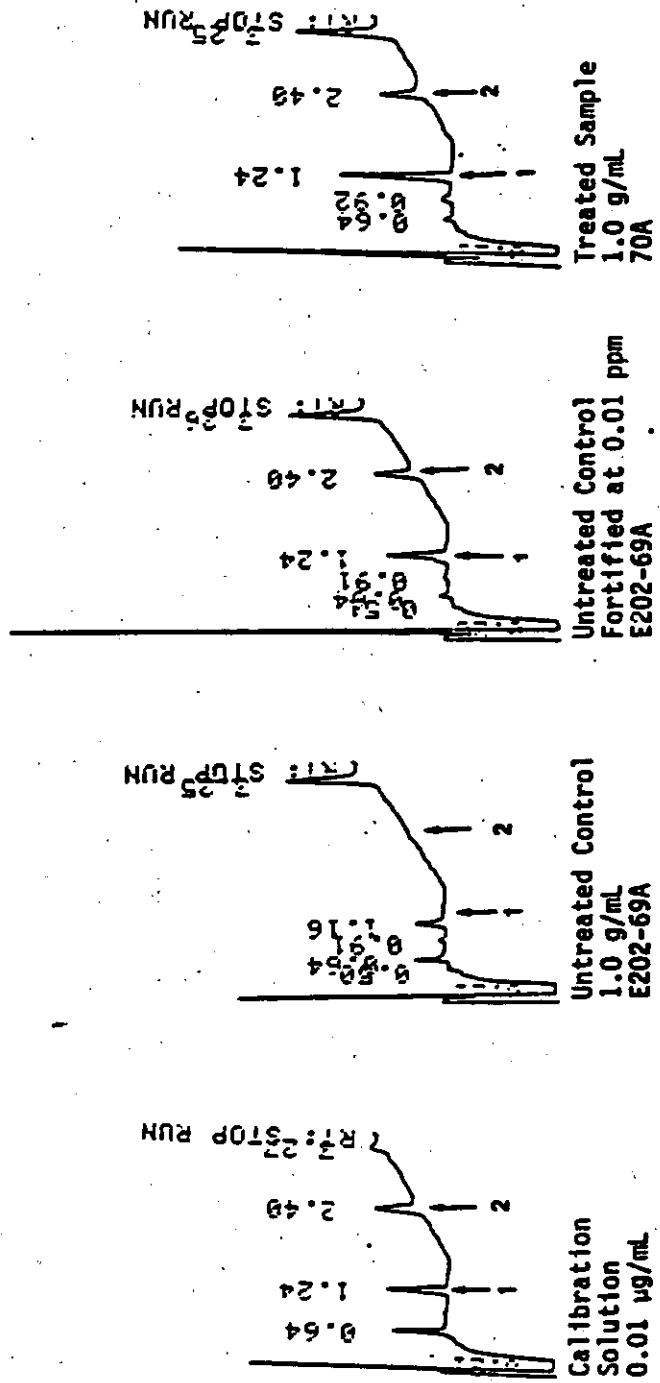


FIGURE F1. EXAMPLE CHROMATOGRAMS FOR THE DETERMINATION OF EPTC (1) AND EPTC SULFOXIDE (2) IN SOIL.

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**APPENDIX F**  
**Example Chromatograms**

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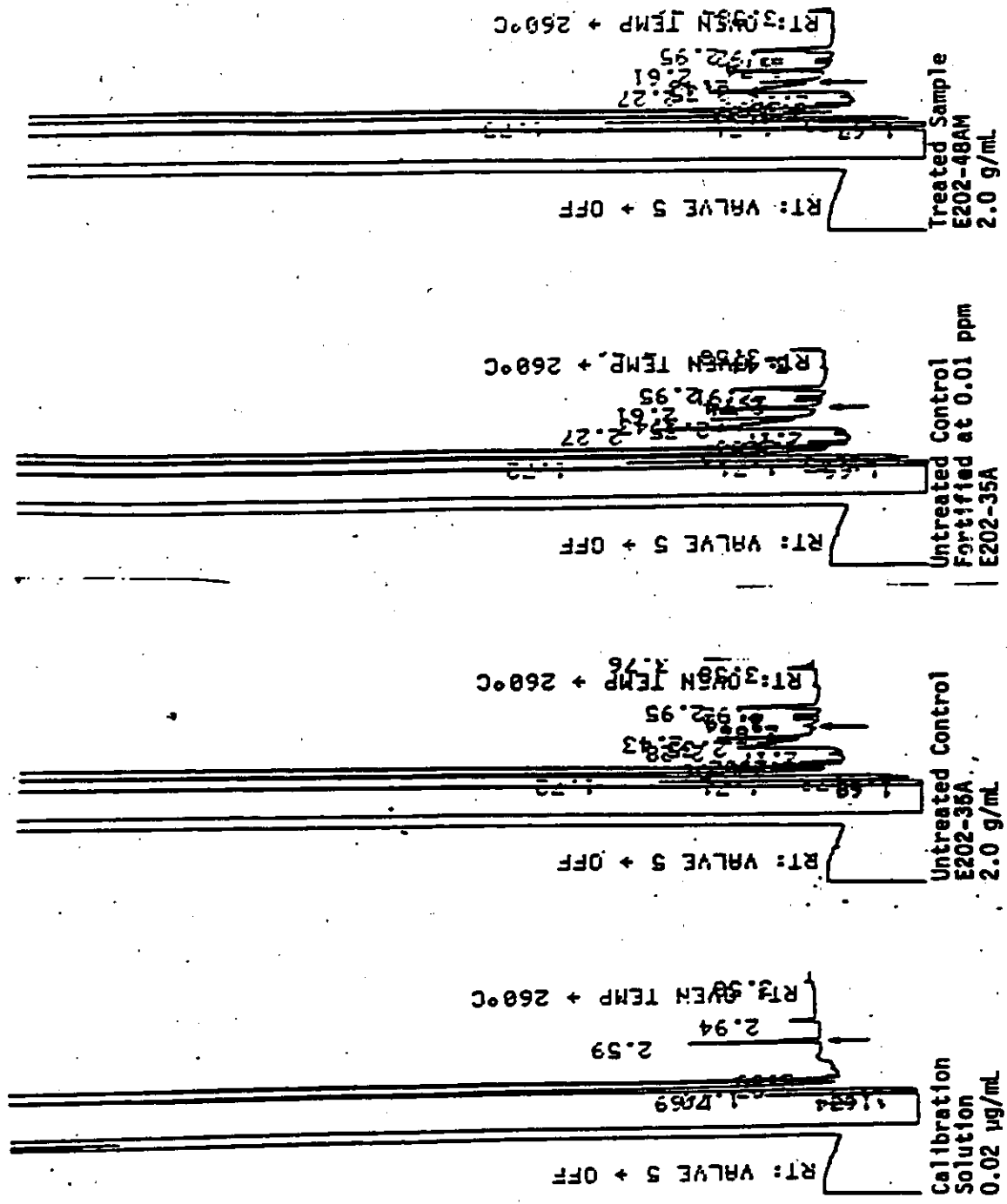


FIGURE F2. EXAMPLE CHROMATOGRAMS FOR THE DETERMINATION OF DI-N-PROPYLAMINE RESIDUES IN SOIL

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**END**